

1. Identification of Substance and Supplier

Product Name	IntelliBond®M	
	Manganese Hydroxychloride	
Alternative Names	Manganese Chloride Hydroxide	
Thermative Publics	Tribasic Manganese Chloride	
	Basic Manganese Chloride	
Recommended Use	Animal feed additive	
of Chemical	Animal feed additive	
Use Restrictions	IntelliBond®M is intended only for use as a source of manganese in animal feeds or research	
	purposes.	
	Micronutrients	
Manufacturer's	1550 Research Way	
Information	Indianapolis, Indiana 46231	
	317-486-5880	
Emergency Phone	<u>CHEMTREC</u> (800)424-9300	
Number	Micronutrients (317) 486-5880	
2. Hazards Identification		

GHS Classification of Substance National or Regional Information	Category 5 Acute Toxicity Category 3 Skin Irritant Category 2B Eye Irritant Category 2 Target Organ Toxicity from Single Exposure Not Applicable
GHS Label Elements	WARNING May be harmful if swallowed Causes mild skin irritation Causes eye irritation May cause damage to organs (red blood cells, liver, pancreas, and lung cells) if inhaled.
Other Hazards	None known

3. Composition / Information on Ingredients

Ingredient Name	CAS Number	EC Number	Percent of Total Weight
Manganese Hydroxychloride (Mn2(OH)3Cl)	39438-40-9	Not Applicable	>95%
Inert Ingredients	Not Applicable	Not Applicable	<5%

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4. First Aid Measures

Eye	Flush eyes with large amounts of water for at least 15 minutes. If irritation persists, seek medical	
	advice.	
Skin	Wash exposed skin with soap and water. If irritation persists, seek medical advice.	
Ingestion	Contact Poison Control and occupational physician.	
Inhalation	Remove individual to fresh air, and seek medical advice.	
Note to Physician	Symptoms of acute manganese exposure include lung irritation.	
	Treat symptomatically.	
5. Firefighting Measures		

Suitable extinguishing media	Utilize compatible fire extinguishing media, including water and any dry media, carbon dioxide (CO_2)
Fire and Explosion Hazards	Material is not considered combustible. Material may melt with decomposition under fire conditions.
PPE and precautions for firefighters	Self-contained breathing apparatus may be appropriate when fighting fires with manganese compounds present.

6. Accidental Release Measures		
Suggested PPE, Equipment and Procedures	Wear disposable coveralls, FFP2 / P2 filter mask, rubber gloves, and protective eye goggles or total face protection.	
Environmental Precautions	Prevent the product from entering water courses or sewers.	
Methods and materials for containment and cleanup	Material is dry powder form. Lightly sweep or vacuum material to collect. Place in a clean, dry container.	

7. Handling and Storage

	8 8	
Handling Precautions	Store in a cool, dry place. Practice good personal hygiene when handling product. Avoid dust formation. Do not breathe dust. Handle in a well-ventilated area or wear adequate respiratory protection (FFP2/P2 filter mask). Avoid contact with skin and eyes using working clothes, gloves and protective glasses. Do not eat, smoke or drink during use. After use keep the packaging tightly closed.	
Storage Precautions	Do not allow bags to become wet, or exposed to fire or extreme heat. Keep in sealed containers away from humidity and sunlight. Store the product in a well-ventilated warehouse away from flammable products. Keep out of reach of children, animals and unauthorized personnel.	
8 Exposure Controls / Personal Protection		

	8. Exposure Controls / Personal Protection
Occupational Exposure Limit Values	There are no TLV established specifically for manganese hydroxychloride. The values provided are for Elemental Manganese. OSHA 8 hr PEL – 1 mg/m ³ NIOSH 15 min STEL – 3 mg/m ³
Engineering Controls	Local or general area ventilation to control dust.
Individual	Protective eyewear is prudent, especially in dusty areas
Protection	Practice good personal hygiene when handling materials.
Measures	Respiratory protection should be selected appropriate to the dustiness of the work environment

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9. Physical and Chemical Properties

Appearance	Brown particulates (typical particle size 20 - 300 µm)
Odor	Odorless
Odor Threshold	Not applicable
pH	6.0 – 7.5 in water, measured by EPA method SW846-9045
Melting Point / Freezing Point	Melting Point – 617°F
Initial Boiling Point and Boiling Range	Not Applicable
Flash Point	Not Applicable
Evaporation Rate	Not Applicable
Flammability	Non-Flammable
Upper / Lower flammability or explosive limits	Not Applicable
Vapor Pressure	Not Applicable
Vapor Density	Not Applicable
Relative Density	2.8 - 3.0
Solubility	Material is soluble in mineral acids. Material soluble in ammonia, amine and EDTA solutions under complex formulation.
Partition Coefficient; n- octanol / water	Not Applicable
Auto-Ignition Temperature	Not Applicable
Decomposition Temperature	617°F

10. Stability and Reactivity

Chemical Stability	Stable
Possibility of	
Hazardous	Will not occur
Reactions	
Conditions to	None Known
Avoid	None Known
Incompatible	Strong oxidizing agents, organic peroxides, strong acids.
Materials	Strong oxidizing agents, organic peroxides, strong acids.
Hazardous	Will decompose when heated about 617°F, May decompose to produce toxic fumes of
Decomposition	
Products	manganese chloride.

11. Toxicological Information

This material was subjected to a research study involving feeding this material to anim	
	ials in
Toxicological varying concentrations greater than normal animal feed additive concentrations. The resu	ults of
characteristics the study indicate that the animals were able to substitute this manganese material for	or the
and symptoms manganese supplement that they had been accustomed to being fed with no adverse l	health
effects.	
Delayed Effects None known	
Immediate Ingestion of large amounts of manganese containing material may be harmful.	



Effects	Metal chloride compounds have been reported to cause eye and / or skin irritation, which may be	
	an allergic reaction. Discoloration of skin may occur, but it is not indicative of injury or illness.	
	Manganese compounds can be toxic by ingestion.	
	Manganese compounds may be toxic if inhaled. Inhalation of metal dust has caused damage to	
	red blood cells, liver, pancreas, lung cells, and has the potential to cause neurological effects.	
Chronic Effects	May cause potential neurological effects.	
Acute Toxicity	Elemental Manganese - Oral LD ₅₀ (rat) 9,000 mg/kg.	
Estimates	LD ₅₀ has not been established for this product.	

12. Ecological Information		
Ecotoxicity	None known	
Persistence and	The product is not environmentally persistent. It will release manganese as a trace mineral	
degradability	when it reacts with acids, bases, or complexing reagents.	
Bio-accumulative	Manganese is an essential trace mineral, which is needed to sustain normal metabolic functions.	
potential	Manganese is not bio-accumulative, and is readily cleared and excreted.	
Mobility in soil	Not Applicable	
Other adverse effects	Not Applicable	
13. Disposal Considerations		

15: Disposal Considerations		
Description of	Waste residues are not anticipated outside of commercial packaging or unintended spills of	
waste residues	material.	
Safe Handling		
and Disposal	Dispose of contents/containers in accordance with local/regional/international regulations.	
methods		
14 There are art Information		

14. Transport Information		
UN Number	Material is not regulated by DOT/ADR	
UN Proper Shipping Name	Material is not regulated by DOT/ADR	
Transport Hazard Class(es)	Material is not regulated by DOT/ADR	
Packing Group	Material is not regulated by DOT/ADR	
Marine Pollutant	No	
Special Precautions	Not Applicable	

15. Regulatory Information		
Applicable Regulations	 Manganese compounds are considered to be toxic chemicals subject to the reporting requirements of Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR 372. SARA – Acute Health Hazard Tier I / Tier II (40 CFR 370.25) reporting required if present and on-site in quantities equal to or exceeding 10,000 pounds. SARA Title III – Section 313 Form R / TRI Reportable Chemical (Manganese Compounds) 	
16. Other		
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	determination as to the suitability of the information for their particular purpose(s).
SDS preparation	Steve Lucas, Carla Jackson
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